

Lead Poisoning: A Heavy (Metal) Topic of Discussion

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Spring 2026 CLPC Series

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Objectives

1. Describe the pathophysiology of lead poisoning.
2. Discuss clinical manifestations of lead toxicity.
3. Explain the role of the clinical laboratory in detection and monitoring of lead exposure.

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What is lead?

- Pb, from Latin word *plumbum*
- Atomic number 82
- Soft, malleable bluish-white metal
- Heavy metal: large atomic mass with densely packed atoms

PERIODIC TABLE OF THE ELEMENTS

Legend:

- Alkali Metal
- Alkaline Earth
- Transition Metal
- Rare Metal
- Metalloid
- Nonmetal
- Halogen
- Noble Gas
- Lanthanide
- Actinide

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Lead, cont.

- Occurs naturally in small amounts in the earth's crust
- Found in concentrated ore deposits widely distributed throughout the world: US, Canada, Spain, Germany, South America, Africa, and Australia
- Harmful or toxic if inhaled or swallowed and can pose serious health risks
 - Young children
 - Pregnant females

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Historical context

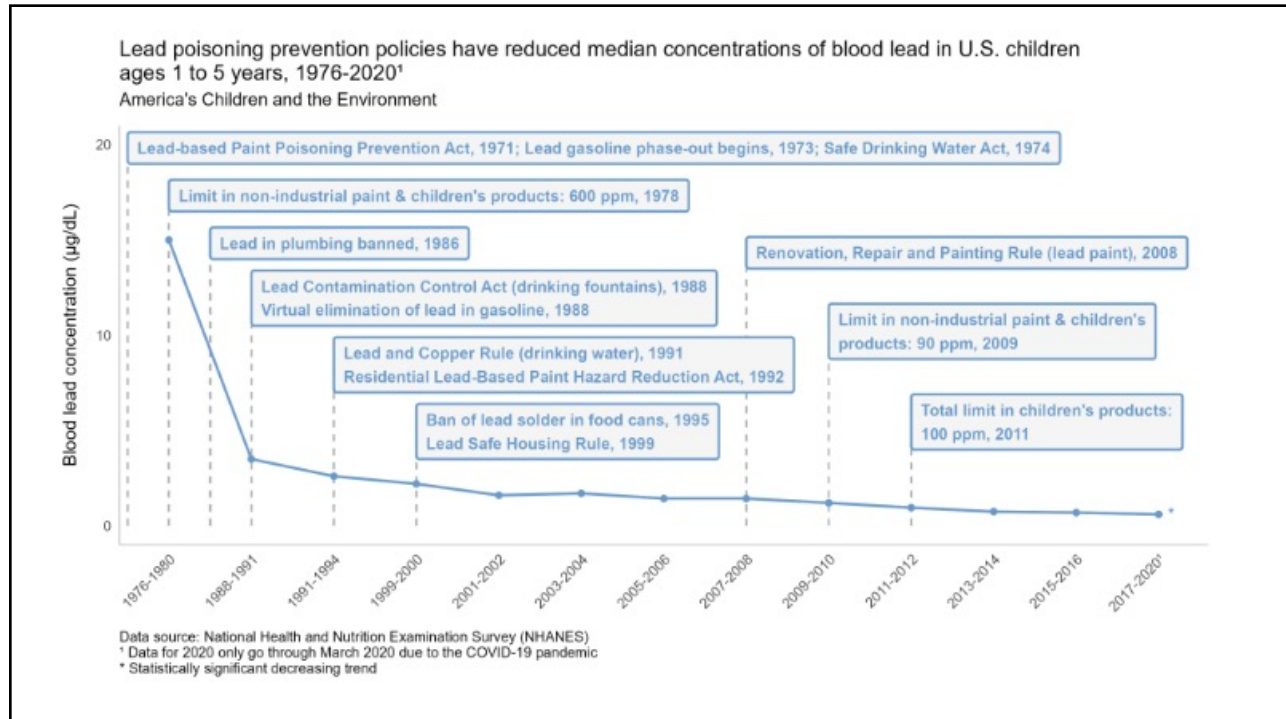
- Mentioned often in early biblical accounts
- Used by Babylonians as plates on which to record inscriptions
- Romans: tablets, water pipes, coins, wine vessels, cooking pots and utensils
- Long associated with toxicity and illness
- Benjamin Franklin described lead poisoning in early printers (1763)

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Modern era (in the U.S.)

- Became a major health concern in the 20th century
 - Leaded gasoline phased out starting in 1973; officially banned in 1996
 - Lead paint banned for residential use in 1978
 - Lead in plumbing banned in 1986
- Remains a significant public health issue despite regulations

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Sources of Lead Exposure

- **Environmental:** lead-based paint, contaminated soil (“legacy contamination”), drinking water
- **Consumer products**
 - Imported: toys, costume jewelry, cosmetics, vinyl lunchboxes & backpacks, foods, spices, or candy
 - Ammunition and fishing tackle: lead bullets and shot, lead sinkers
 - Hobby materials: stained glass work, pottery glazes, oil painting

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Sources, cont.

- **Occupational:** battery manufacturing & recycling, construction & renovation, auto repair, firing ranges, mining, ship building & bridge construction, stained glass and pottery work
 - Take-home contamination
- **Industrial contamination:** lead refineries, battery recycling facilities, mining operations; air/water/soil contamination from industrial sites

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Benefits of lead

- Paint: superior coverage, faster drying, durability, weather resistance, rust inhibition
- Gasoline: more powerful and efficient engines
- Batteries: efficiently stores and releases electricity
- Radiation shielding: blocks X-rays and gamma rays
- Construction and plumbing: durable and resists corrosion
- Ammunition: increased mass in small volume, improved accuracy
- Glass: increased weight, sparkle, and ease of cutting

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<h2>Vulnerable Populations</h2> <ul style="list-style-type: none">• Children under 6 y/o• Pregnant individuals• Low-income populations• Residents in older housing• Those consuming imported foods/spices• Certain occupations	
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<h2>Clinical Symptoms of Lead Toxicity</h2>	
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Acute Exposure (rare)

- Occurs with very high exposure (BLL >70 µg/dL)
 - Severe abdominal pain (“lead colic”)
 - Vomiting
 - Constipation
 - Seizures
 - Encephalopathy
 - Coma
 - Death (if untreated)

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Acute Exposure, cont.

- Most commonly from:
 - Industrial accidents
 - Ingestion of large amounts of lead-containing materials
 - Criminal poisoning (rare)

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Chronic Exposure (more common)

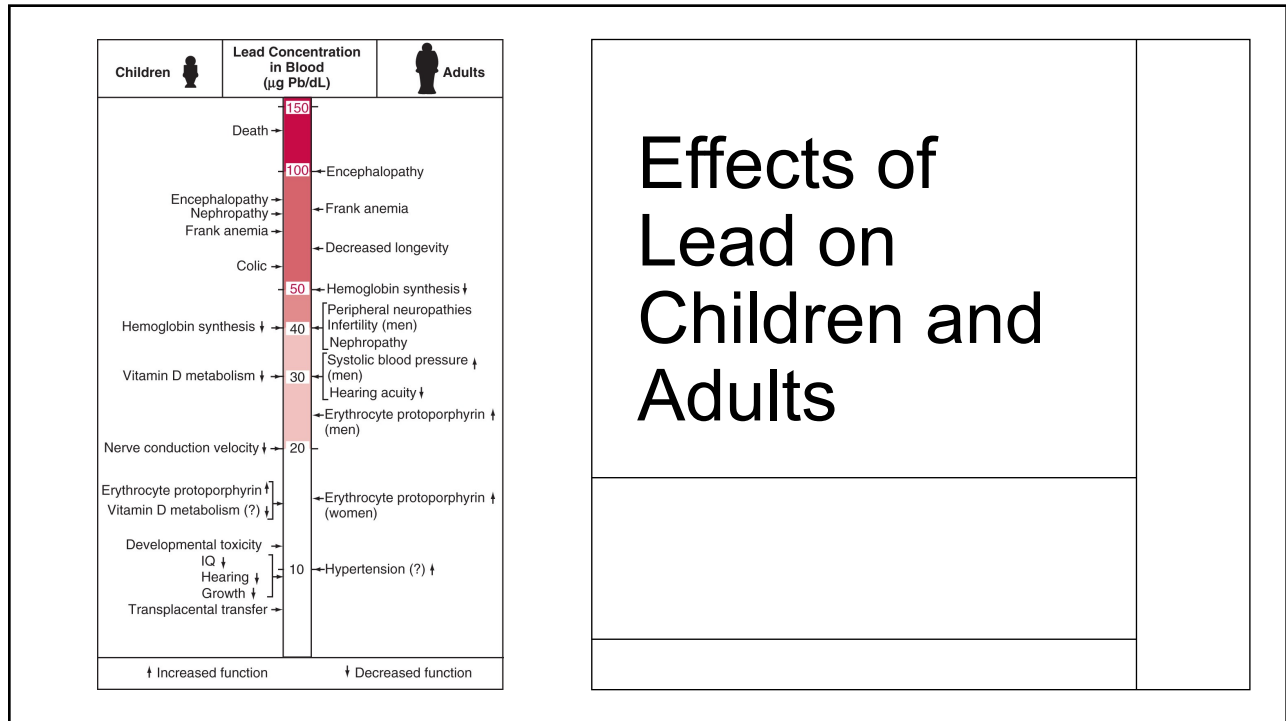
- Symptoms develop gradually due to continued exposure
- Often non-specific
- May be initially attributed to other causes
- Children particularly vulnerable

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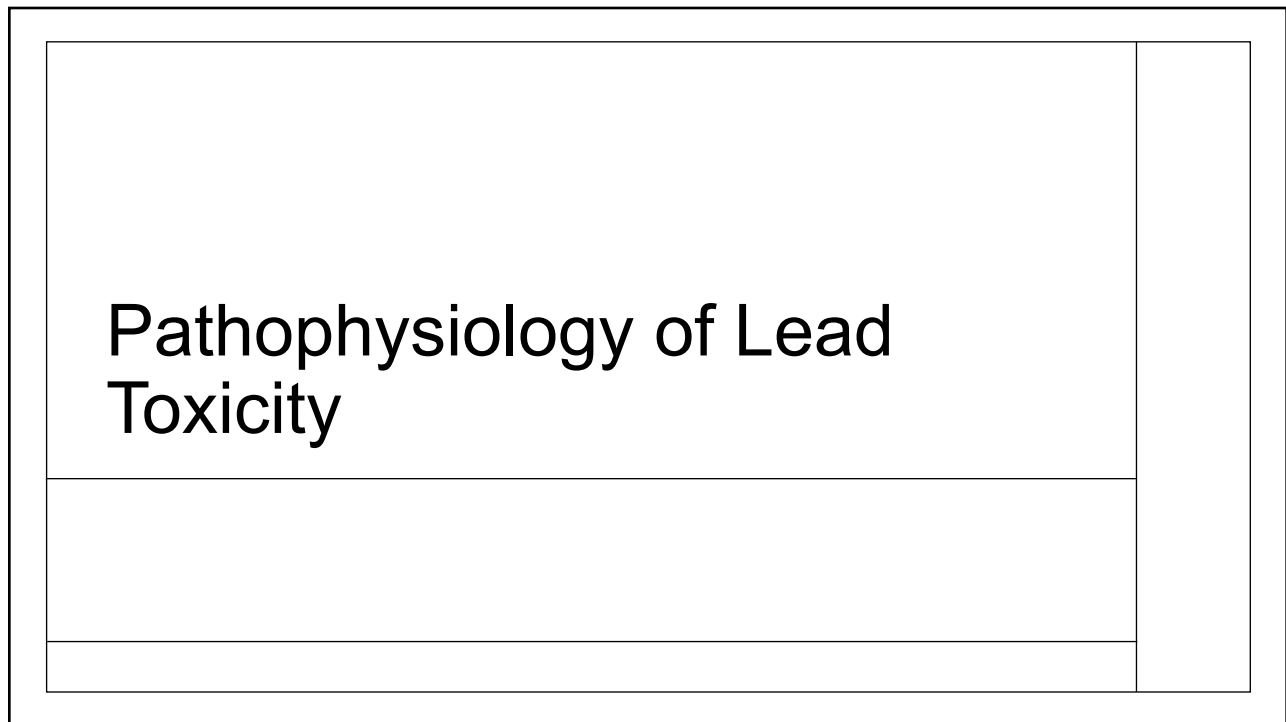
Physiologic Effects

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|--|--|
| <ul style="list-style-type: none">• Neurologic• Hematologic• Renal | <ul style="list-style-type: none">• Gastrointestinal• Cardiovascular• Reproductive |
|--|--|

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Routes of Lead Absorption

- **Gastrointestinal (GI)**
 - Children: ~50% absorption
 - Adults: 10-15% absorption
 - Enhanced by empty stomach, iron and/or calcium deficiency
- **Respiratory (most efficient):** nearly 100% of inhaled lead particles <2.5 μm
- **Dermal:** minimal

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Lead Distribution in the Body

- | | |
|---|---|
| <ul style="list-style-type: none"> • Blood (half-life ~30 days) <ul style="list-style-type: none"> • ~99% bound to RBCs via hemoglobin • <1% in plasma • Blood lead level (BLL) reflects recent exposure • Soft tissues (half-life ~40 days) <ul style="list-style-type: none"> • Liver, kidneys, brain, muscles • Sites of toxic effects | <ul style="list-style-type: none"> • Bone (half-life 20-30 years) <ul style="list-style-type: none"> • ~95% of total body burden in adults • ~70% in children • Mobilized during: <ul style="list-style-type: none"> • Pregnancy • Lactation • Menopause • Hyperthyroidism • Osteoporosis |
|---|---|

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Excretion of Lead

- Renal: ~75%
- Fecal: ~15%
- Minor routes: hair, nails, sweat, breast milk
- Occurs slowly, so bioaccumulation is a concern

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Mechanisms of Lead Toxicity

- **Direct DNA damage**
 - Interference with DNA repair mechanisms
 - Potential carcinogenic effects
- **Calcium interference**
 - Mimics calcium
 - Disrupts release of neurotransmitters
 - Responsible for neurotoxic effects
- **Oxidative stress**
 - Depletes natural antioxidants
 - Damages cell membranes
 - Contributes to shortened RBC survival

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Mechanisms of Lead Toxicity, cont.

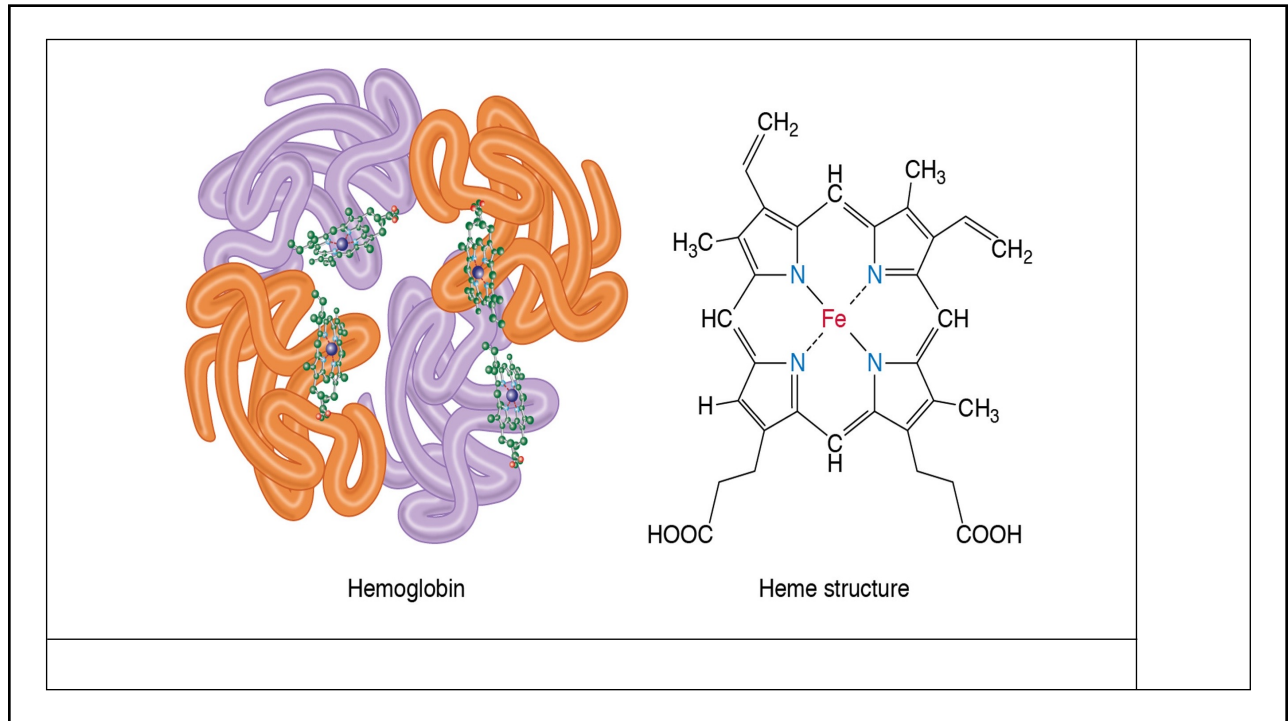
- **Inhibition of RNA degradation in maturing RBCs**
 - RNA aggregates accumulate within the cell
 - Coarse, dark blue granules visible on Wright stain
 - May shorten RBC lifespan

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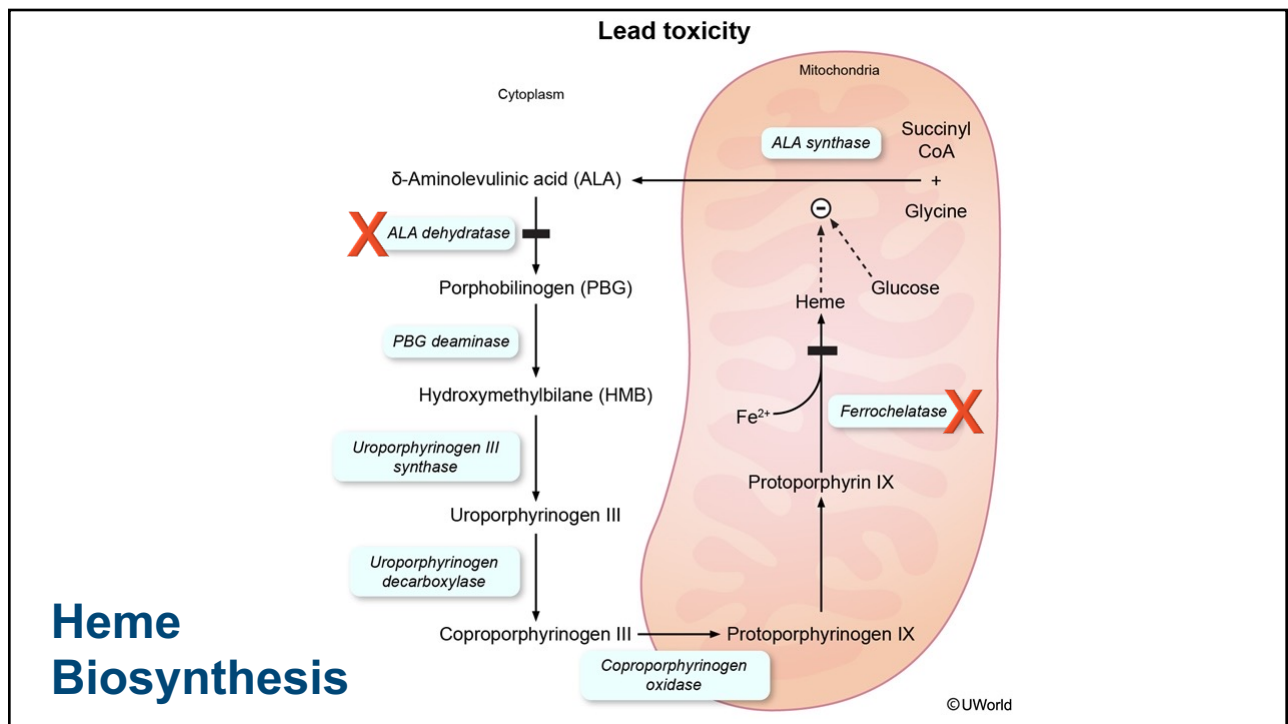
Mechanisms of Lead Toxicity, cont.

- **Disruption of heme synthesis**
 - Inhibits two key enzymes
 - **δ -Aminolevulinic acid (ALA) dehydratase** – most sensitive to lead; results in δ -ALA accumulation in blood & urine
 - **Ferrochelatase** – most clinically significant; results in excess incomplete heme rings in blood and iron accumulation in bone marrow

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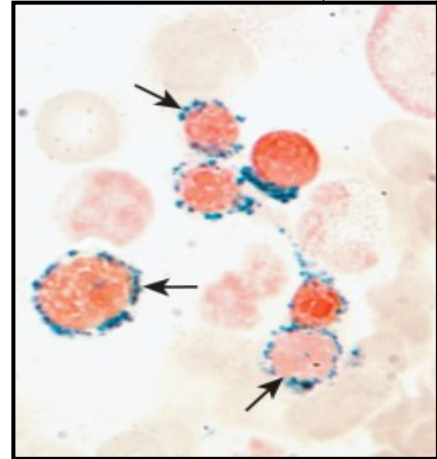


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Hematologic Consequences of Lead Toxicity

- **Sideroblastic anemia**

- Ringed sideroblasts in bone marrow
 - Iron encircles developing RBC nuclei in mitochondria
 - Mitochondria rupture, causing iron overload
- Basophilic stippling in peripheral blood
 - Punctate RNA aggregates in mature RBCs
- Decreased RBC survival leads to anemia
 - Ineffective marrow production of RBCs
 - Defective RBCs have shortened lifespan



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Laboratory Testing and Findings

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Pre-analytical Considerations

- **Contamination prevention is critical**
 - Standard evacuated tubes may contain lead
 - Clean skin thoroughly and use proper venipuncture technique
 - Avoid hemolysis (most circulating lead is intracellular)
 - Capillary samples acceptable for screening only
- Specimens are stable for 30 days if refrigerated

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Blood Lead Level (BLL)

- Gold standard
- Specimen: Whole blood, K₂EDTA – preferably royal blue top (trace element) tube
- Analysis methods: mass spectrometry – highly sensitive and specific; voltammetry – POC devices for screening

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Reference Values

- **No safe blood lead level (BLL) established by CDC**
- CDC currently uses **3.5 µg/dL** as the reference value
- Any detectable BLL means the person has been exposed to lead.
 - Typical BLL in U.S. adults is ~0.85µg/dL or less
- Most experts recommend intervention at levels ≥ 10 µg/dL
- Pregnant women: begin interventions at ≥ 5 µg/dL

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Blood Lead Level (µg/dL)	Interpretation	Recommended Actions	
< 3.5	Within expected range	- Routine screening based on age/risk - Provide education on lead exposure prevention	
3.5 – 9	Elevated (above reference value)	- Confirm with repeat test (venous sample) - Educate caregivers on sources of lead - Nutritional counseling (iron, calcium) - Environmental exposure history	
10 – 19	Moderately elevated	- Report to local health department - Environmental investigation of home - Monitor development and nutrition	
20 – 44	High	- Full medical evaluation - Detailed environmental investigation - Consider abdominal X-ray (if ingestion suspected) - Consult specialists (e.g., toxicology) - Follow-up testing frequently	
45 – 69	Very high	- Urgent medical evaluation - Initiate chelation therapy - Hospitalization may be required - Immediate environmental intervention	
≥ 70	Severe, medical emergency	- Emergency hospitalization - Immediate chelation therapy - Intensive monitoring (neurologic risk) - Rapid environmental control	www.cdc.gov/lead-prevention

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Zinc Protoporphyrin (ZPP)

- AKA Free Erythrocyte Protoporphyrin (FEP)
- When lead inhibits ferrochelatase, zinc substitutes for iron in protoporphyrin ring
- Is easily detected by fluorescence
- ZPP/FEP elevation reflects chronic lead exposure (over ~120 days)

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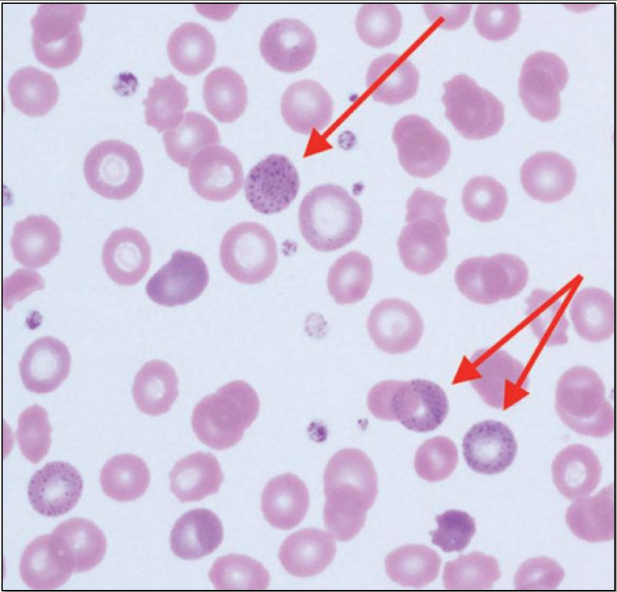
ZPP, cont.

- Normal: <35 µg/dL
- Elevated: indicates chronic lead exposure OR iron deficiency
- Limitations: Not specific for lead, not sensitive at levels <25 µg/dL
- Utility as a screening tool for adults and to monitor chronic exposure – remains elevated longer than BLL once exposure stops

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Hematology Lab Findings	
Complete Blood Count (CBC) <ul style="list-style-type: none">• Mildly decreased Hgb• Decreased MCV, MCH, MCHC• Increased RDW• Increased reticulocyte count	Peripheral Blood Smear <ul style="list-style-type: none">• Microcytic, hypochromic cells• Polychromasia• RBC size variation (anisocytosis)• Basophilic stippling

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Basophilic Stippling in Mature RBCs	
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Iron Studies

Test	Lead Poisoning	Iron Deficiency Anemia (IDA)
Serum Iron	↑ Increased	↓ Decreased
Serum Ferritin	↑ Increased	↓ Decreased
%Transferrin Saturation	↑ Increased	↓ Decreased
Total Iron Binding Capacity	Normal or ↓ decreased	↑ Increased
ZPP/FEP	↑↑ Markedly increased	↑ Increased
Bone Marrow Iron Stores	Normal or ↑; ringed sideroblasts	↓ Absent

Key Distinction: Lead poisoning causes a functional iron deficiency, while iron deficiency anemia is caused by true iron depletion.

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Other Significant Findings

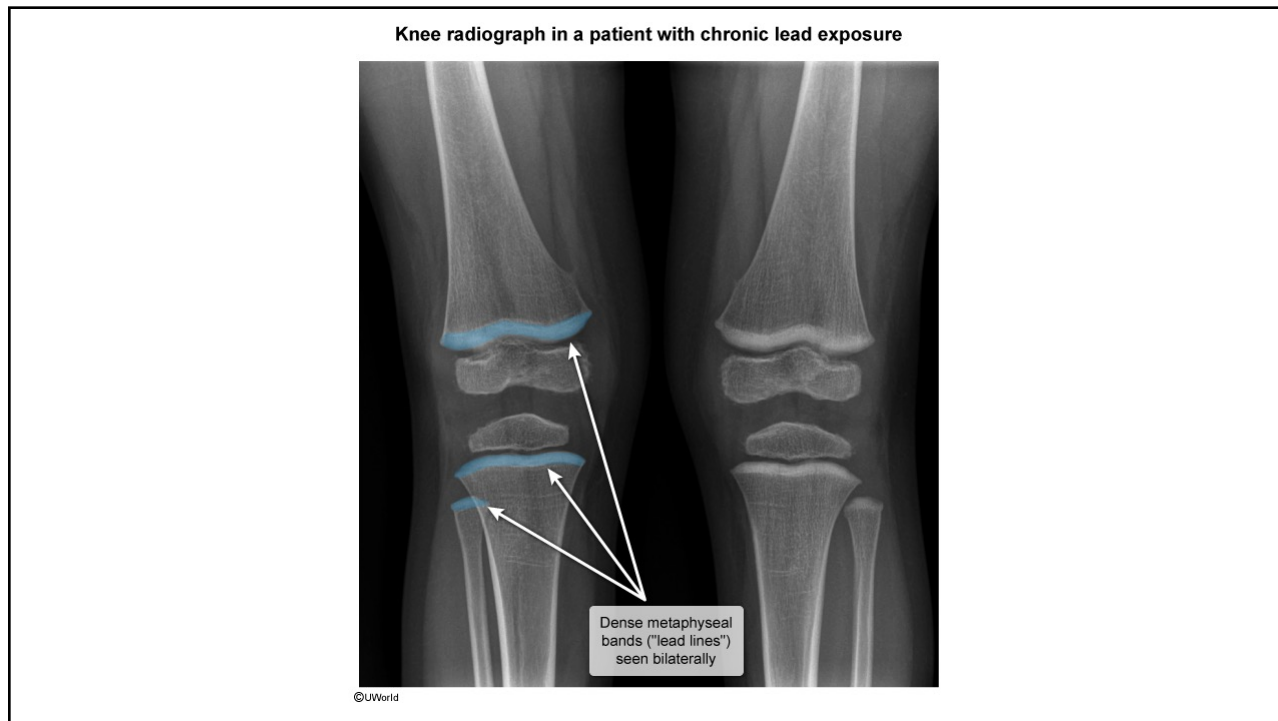
Chemistry

- Elevated uric acid
- Elevated creatinine
- Elevated BUN

Imaging/Visual

- X-ray
 - “Lead lines” on the ends of bones in children
 - Possible lead particles seen in abdomen if recently ingested
- Dental
 - Burton’s line
 - Severe enamel defects

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<h1>Management and Treatment</h1>		

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<h2>Remove Source (most important)</h2>		
<ul style="list-style-type: none">• For children<ul style="list-style-type: none">• Home inspection by certified lead inspector• Cleaning, repairs, removal of lead hazards• Relocation if necessary	<ul style="list-style-type: none">• For workers<ul style="list-style-type: none">• Engineering controls, PPE• Hygiene practices• Medical removal from job if necessary	

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Nutritional Interventions

- Iron supplementation – deficiency increases lead absorption
- Calcium supplementation – lead competes with calcium for absorption
- Vitamin C – may enhance lead excretion
- Regular meals – fasting increases lead absorption
- Balanced diet – adequate protein, limited fat

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Chelation Therapy

- Binds lead in blood or soft tissues, forms compounds that can be safely excreted in urine or feces
- May require multiple courses if BLL remains elevated
- Rebound phenomenon: BLL may increase after chelation as lead redistributes from bone
- In children: BLL ≥ 45 $\mu\text{g}/\text{dL}$ OR if symptomatic, or BLL continue to rise despite intervention
- In adults: BLL $\geq 80-100$ $\mu\text{g}/\text{dL}$ OR if symptomatic regardless of BLL

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Chelation Controversy

- Benefit of chelation at lower BLL (20-44 $\mu\text{g}/\text{dL}$) unclear
- Studies in lead-exposed children lowered BLL but did not improve cognitive impairments
- Current practice is to reserve chelation for higher BLL or symptomatic cases

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Regulatory Oversight

- Centers for Disease Control and Prevention (CDC)
- Environmental Protection Agency (EPA)
- Occupational Safety and Health Administration (OSHA)
- Food and Drug Administration (FDA)
- Consumer Product Safety Commission (CPSC)
- Department of Housing and Urban Development (HUD)
- Louisiana Department of Health (LDH)

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Key Takeaways

- Lead poisoning remains a significant public health problem despite declining prevalence.
- No safe blood lead level has been identified.
- Blood lead level (BLL) is the gold standard test; use trace element tubes to avoid contamination.
- Heme synthesis disruption is the key pathophysiological mechanism affecting laboratory findings.
 - Zinc protoporphyrin (ZPP) reflects chronic exposure but is not specific for lead.
 - Differentiate lead toxicity from iron deficiency using iron studies.
- Children are most vulnerable to neurotoxic effects.
- Environmental remediation is the cornerstone of treatment.
- Chelation therapy is reserved for higher BLL or symptomatic patients.
- Laboratory professionals play a vital role in screening, diagnosis, and public health surveillance.

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- <https://www.osha.gov/lead>
- <https://ldh.la.gov/lead-poisoning-prevention>

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Thanks for listening!
Any questions?

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